

Correlative Atomic Coordination and Interfacial Charge Polarity in Al₂O₃/GaN and Al₂O₃/Si Heterostructures

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 Al_2O_3 is used to form Al_2O_3/GaN and Al_2O_3/Si heterostructures for many electronic and optoelectronic devices. Yet the origins of the positive interfacial charges in the Al_2O_3/GaN heterostructure and negative interfacial charges in the Al_2O_3/Si heterostructure are often elusive. Herein, in-depth studies of the Al_2O_3/GaN and Al_2O_3/Si heterostructures are conducted, especially on how the atomic coordination structure affects the interfacial charges. It is discovered that the octahedral $[AlO_6]^9$ — coordination surpassed tetrahedral $[AlO_4]^5$ — coordination at the Al_2O_3/GaN interface, whereas tetrahedral $[AlO_4]^5$ — coordination dominated at the Al_2O_3/Si interface. Therefore, the interfacial charge polarity is correlated with the nonstoichiometry atomic coordination of Al_2O_3 at the two interfaces. This study reveals the atomic origin of charge polarity at high-k/semiconductor interfaces and could facilitate the development of high-performance optoelectronic and electronic devices through engineering the atomic coordination structure of the high-k materials.

1. Introduction

High-k dielectrics are widely used in electronic and optoelectronic devices to passivate the dangling bonds at the semiconductor surfaces or as gate insulators to reduce the gate leakage current. However, substantial interfacial charges that occur at high-k dielectric/semiconductor interfaces can severely impact device design and engineering, and the charge origins are often elusive. For example, Al_2O_3 has been adopted in many crucial devices including GaN high-electron-mobility transistors (HEMTs) and Si solar cells. In Si solar cells, Al_2O_3 is used as the surface passivation layer owning to high-density negative charges occur the Al_2O_3 /Si interface. These negative charges can shield the electrons from the Si surface thus mitigating the surface recombination rate. Structural impurities in Al_2O_3 such as Al vacancies, O dangling bonds, O interstitials, and H interstitials are proposed to be the origin of the negative charges.

Al₂O₃ has also been widely used in GaN HEMTs as the gate dielectric to decrease the gate leakage current and increase

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the gate breakdown voltage.[10-12] By simply extending the gate dielectric thickness, enhancement-model (E-mode) GaN HEMTs can be achieved if the charges at the Al₂O₃/GaN interface are also negative.[12] Nevertheless, positive charges that occur at the Al₂O₃/GaN interface can negatively shift the threshold voltage (V_{TH}) , retarding the achievement of E-mode GaN HEMTs.^[13–15] During the atomic layer deposition (ALD) process, the incomplete oxidation of trimethylamine precursors will result in residual carbon impurities in Al₂O₃, which is proposed as the origin of the positive interfacial charges of the Al₂O₃/GaN heterostructure. [16,17] Additionally, Al interstitials and O vacancies in Al₂O₃ are also regarded as the sources of the positive interfacial charges in Al_2O_3 . [9,18] Moreover, the

polarity of GaN (Ga-polar and N-polar) has negligible influence on the positive interfacial charges occur at the ${\rm Al_2O_3/GaN}$ heterostructure. [19]

The flat-band voltage (V_{FB}) dependence on the gate dielectric thickness of Al₂O₃/GaN and Al₂O₃/Si metaloxide-semiconductor (MOS) capacitors is shown in Figure 1a,b, respectively, where the V_{FB} was derived from the capacitance-voltage (C-V) characteristics. [19,20] The $V_{\rm FR}$ decreases linearly with the increased Al2O3 thickness of the Al₂O₃/GaN MOS capacitor, whereas, it increases linearly with the increased Al₂O₃ thickness of the Al₂O₃/Si MOS capacitor, indicating positive and negative charges occur at the Al₂O₃/GaN and Al₂O₃/Si interfaces, respectively. This result also implies that the charges which can affect the $V_{\rm FB}$ of the devices are not intrinsic to ALD Al₂O₃; instead, semiconductor surfaces can play an important role in determining the polarity of the interfacial charges. In addition, as shown in Figure 1a,b (dotted line), positive charges can occur at other high-k/GaN interfaces, such as NiO/GaN, HfO2/GaN, and CeO2/GaN. [20,21] Similarly, the negative charges also occur at the interfaces between these high-k dielectrics and Si.[20,21] Nonstoichiometry atomic ratios of Al_2O_3 were found in the Al_2O_3/GaN and Al_2O_3/Si heterostructures where Al and O dominate at the Al₂O₃/GaN and Al₂O₃/Si interfaces, respectively. [1-3,5,6,22] In this work, we found the atomic coordination structure of Al₂O₃ was also different on the GaN and Si surfaces which could explain the different interfacial charge polarity of the Al₂O₃/GaN and Al₂O₃/Si heterostructures. Moreover, this mechanism could possibly be adopted to explain the interfacial charges that occur at other high-k/semiconductor interfaces.

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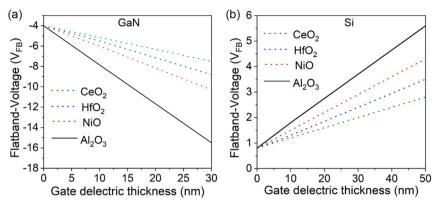


Figure 1. a) Calculated V_{FB} of: a) GaN and b) Si metal-oxide-semiconductor (MOS) capacitors as a function of the gate oxide thickness. [19,20]

2. Results and Discussion

The *C*–*V* technique was commonly used to characterize the fixed charge and trap states at the Al₂O₃/GaN and Al₂O₃/Si interfaces. The results of C-V measurements depend on the electric field distribution at the interfaces of the Al₂O₃/GaN and Al₂O₃/Si heterostructures. However, the introduction of the metal gate contact in the MOS capacitors could change the electric field distribution at the interfaces. In contrast, KPFM can provide a noninvasive and nondestructive method for mapping the surface potential of different materials.^[23–25] Therefore, in this study, we used KPFM to evaluate the charge polarity and density at the interfaces of the Al₂O₃/GaN and Al₂O₃/Si heterostructures. After coating Al₂O₃ onto the GaN and Si samples, the electric field induced by the charges at the interfaces of the Al₂O₃/GaN and Al₂O₃/Si heterostructures can lead to surface band bending; therefore, the changing surface potential of the bare GaN and Si samples can be probed without introducing the metal gate contact in the Al₂O₃/GaN and Al₂O₃/Si heterostructures using KPFM.

Figure 2 illustrates that the mean potential was reduced from 993 to 661 mV after the bare, clean GaN samples were coated with a thin Al₂O₃ layer; however, the mean potential increased from 305 to 737 mV after the bare, clean Si samples were coated with a thin Al₂O₃ layer. In addition, after coating Al₂O₃, the root mean square roughness increased from 0.57 to 0.63 nm and from 0.48 to 0.75 nm for GaN and Si, respectively, implying the high-quality of ALD Al₂O₃ layers. This indicates that GaN and Si samples have different potential responses after they are covered with a thin Al₂O₃ layer (having opposite interfacial charges). This phenomenon also proves that positive charges at the Al₂O₃/GaN interface and negative charges at the Al₂O₃/Si interface were introduced after Al₂O₃ deposition. The charge densities σ at the Al₂O₃/GaN and Al₂O₃/Si interfaces can be estimated using the Gauss theorem shown in equation^[25]

$$\sigma = \frac{\Delta V_{Al_2O_3} \varepsilon_{Al_2O_3} \varepsilon_0}{de} \tag{1}$$

where $\Delta V_{\text{Al}_2\text{O}_3}$ is the contact potential difference; $\varepsilon_{\text{Al}_2\text{O}_3}$ is the dielectric constant of Al₂O₃; d is the thickness of Al₂O₃; e is the elementary charge. Assuming that all the charges are located

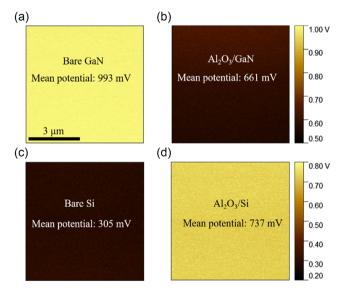


Figure 2. Kelvin probe force microscopy (KPFM) images of the: a) bare, clean GaN samples, b) Al₂O₃/GaN, c) bare, clean Si samples, and d) Al₂O₃/Si.

at the Al₂O₃/GaN and Al₂O₃/Si interfaces, and $\varepsilon_{\text{Al}_2\text{O}_3}=4$ for the ultrathin Al₂O₃. [26] The charges at the Al₂O₃/GaN and Al₂O₃/Si interfaces are $+4.5 \times 10^{12}$ and -5.7×10^{12} cm⁻², respectively. The interfacial charges obtained by the KPFM measurement are close to the values measured by the C-V technique.^[1,11]

Electron energy loss spectroscopy (EELS) is capable of fingerprinting different coordinated forms of the same element. Amorphous Al₂O₃ contains both tetrahedral [AlO₄]⁵⁻ and octahedral [AlO₆]⁹⁻ coordination; therefore, the tetrahedrally and octahedrally coordinated Al with the energy loss of 2.4 eV in separate can be mapped across the interfacial regions of the Al₂O₃/GaN and Al₂O₃/Si heterostructures by EELS measurements. [27-31] EELS has been applied to characterize the Al₂O₃/Si heterostructure, and the tetrahedral $[AlO_4]^{5-}$ coordination was found to dominate at the interface. [27–29,32] This study employs the identical method to measure the Al coordination at the Al₂O₃/GaN interface. Figure 3a,b shows the STEM images of the Al₂O₃/GaN and Al₂O₃/Si heterostructures, respectively,

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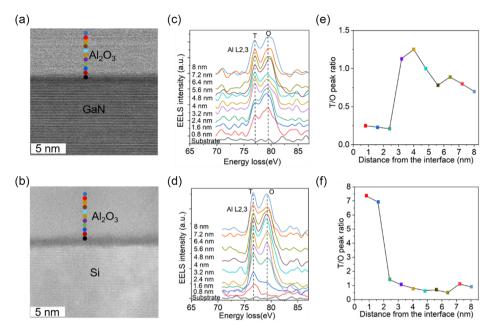


Figure 3. Cross-sectional scanning transmission electron microscope (STEM) images of the: a) Al_2O_3/GaN and b) Al_2O_3/Si heterostructures. The spatially resolved electron energy loss spectroscopy (EELS) spectra of Al L2, 3 edge across the interfacial regions from the: c) GaN and d) Si substrates to the Al_2O_3 . Ratios of the tetrahedral (T) and octahedral (T) and octahedral (T) peak intensity as a function of the distance from the interfaces of the: e) Al_2O_3/GaN and f) Al_2O_3/Si heterostructures to the Al_2O_3 .

and the surface oxides of GaN and Si were observed at interfaces. Figure 3c,d shows the spatially resolved EELS spectra of Al L2, 3 edge across the interfacial regions from the GaN and Si substrates to the Al₂O₃ (the interval of each spectrum is 0.8 nm in distance). The black spectrum that represents the EELS from the native oxides of GaN and Si shows a negligible Al L2, 3 signal. For the Al₂O₃/GaN heterostructure, when the acquisition position is 0.8 nm away from the interface, the octahedrally coordinated Al peaks at 79.3 eV are clearly evident in the spectrum while the tetrahedrally coordinated Al peaks at 76.9 eV have small intensity. When the sampling position distance is 3.2 nm away from the interface, there occurs a dramatic increase of the tetrahedrally coordinated Al. The EELS spectra across the interfacial region from the Si substrate to the Al₂O₃ layer was also measured. For the Al₂O₃/Si heterostructure, tetrahedrally coordinated Al peaks at 76.9 eV is clearly evident in the spectrum when the acquisition position is at a distance of 0.8 nm from the interface; the octahedrally coordinated Al located at 79.3 eV increases abruptly when the sampling position is at a distance of 2.4 nm from the interface. As shown in Figure 3e,f, a quantitative analysis reveals that the octahedral [AlO₆]⁹⁻ coordination dominates at the Al₂O₃/GaN interface whereas tetrahedral [AlO₄]⁵⁻ coordination dominates at the Al₂O₃/Si interface, and the tetrahedral/octahedral (T/O) peak ratio gradually increases and decreases to ≈ 0.7 for the Al₂O₃/GaN and Al₂O₃/Si heterostructures, respectively. Therefore, the stoichiometric derivation of the atomic coordination from the bulk Al₂O₃ is confirmed at both the Al₂O₃/GaN and Al₂O₃/Si interfaces using the EELS measurements.

At the Al $_2$ O $_3$ /Si interface, stoichiometry deviation of O/Al ratios was widely reported where O significantly suppress Al. $^{[1-3,5,6,22]}$ In addition, at the surface of the Si substrate, Si

in the SiO_x films has a tetrahedral coordination.^[5] The physical connection of tetrahedrally coordinated Si and Al₂O₃ leads to the dominant formation of the tetrahedrally coordinated Al at the Al₂O₃/Si interface. [22,33] For III-nitride materials such as AlN and GaN, octahedrally coordinated O-III-O is formed at the surfaces.^[34] Similarly, octahedrally coordinated O-III-O can also imprint their structures onto the adjacent interfacial Al₂O₃, leading to the dominance of octahedrally coordinated Al at the Al₂O₃/GaN interface. Moreover, as the tetrahedral [AlO₄]⁵⁻ coordination peaks at 76.9 eV are O-rich, [22] nonstoichiometry O/Al ratios found in previous reports can be attributed to the dominant formation of tetrahedral [AlO₄]⁵⁻ coordination at the Al₂O₃/Si interface. At the Al₂O₃/GaN interface, nonstoichiometry atomic ratios of Al₂O₃ were also found where Al significantly suppress O, [1] which can be attributed to the dominance of octahedral [AlO₆]⁹⁻ coordination at the interface. To explore how different atomic coordinations can affect the interfacial charge polarity, DFT calculations were performed on the heterostructures where excess Al and O were introduced at the Al₂O₃/GaN and Al₂O₃/Si interfaces, respectively.

Bader charge analysis was performed to quantitatively analyze the electron accumulation and depletion of each atom in the Al_2O_3/GaN heterostructures without and with the excess Al interstitial and the Al_2O_3/Si heterostructures without and with the excess O interstitial. **Figure 4**a,b shows the quantitative results of the charge change of each atom of the Al_2O_3/GaN heterostructures without and with one Al interstitial, respectively. Figure 4c,d shows the quantitative results of the charge change of each atom of the Al_2O_3/Si heterostructures without and with one O interstitial, respectively. The Al interstitial of the Al_2O_3/GaN heterostructure and the O interstitial of the Al_2O_3/Si heterostructure are indicated by arrows in Figure 4b,



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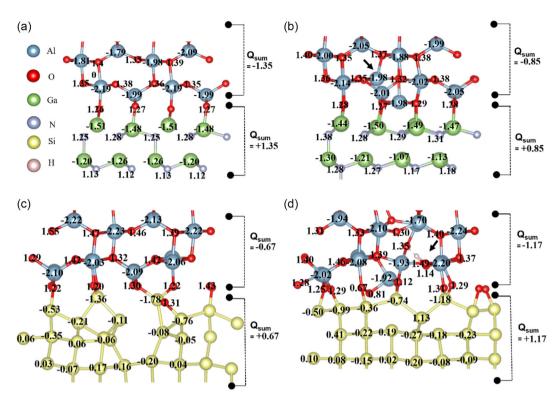


Figure 4. Atomic Bader charges of the Al_2O_3/GaN heterostructures: a) without and b) with one Al interstitial. Atomic Bader charges of the Al_2O_3/GaN heterostructures: c) without and d) with one O interstitial. The gain and loss of electron density on each atom are indicated by the (+) and (-) signs, respectively.

d, respectively. Similar to the Al and O atoms in the lattice sites, the Al and O interstitials have an electron loss of 1.98 e and an electron gain of 1.49 e, respectively. The sum of Bader atomic charges (Q_{sum}) is also shown in Figure 4a–d. The electron gain of Al₂O₃ in the Al₂O₃/GaN heterostructures will be reduced from 1.35 to 0.85 e when the Al interstitial exists at the interface, indicating the Al interstitial will deplete electrons and induce positive interfacial charges. On the contrary, at the interface of the Al₂O₃/ Si heterostructures, the electron gain increases from 0.67 to 1.17 e after interposing the O interstitial, indicating the O interstitial will accumulate electrons and induce negative interfacial charges. Therefore, atomic-coordination-induced nonstoichiometry atomic ratios at the Al₂O₃/GaN and Al₂O₃/Si interfaces are correlated to the interfacial charge polarity by DFT calculations. Indeed, Lucovsky postulated that Al₂O₃ comprises both negatively charged tetrahedrally and positively charged octahedrally coordinated units and different atomic coordination should maintain a ratio to ensure charge neutrality. [5,35] Therefore, the positive interfacial charges of the Al₂O₃/GaN heterostructure and the negative interfacial charges at the Al₂O₃/Si heterostructure can be attributed to the dominance of octahedrally and tetrahedrally units at the Al₂O₃/GaN and Al₂O₃/Si interfaces, respectively.

3. Conclusion

In summary, different interfacial charge polarities occurring at the ${\rm Al_2O_3/GaN}$ and ${\rm Al_2O_3/Si}$ heterostructures were correlated

with the atomic coordination and nonstoichiometry atomic ratios of the interfacial Al_2O_3 . Moreover, we suppose this phenomenon can also be applied to explain the charges that occur at the interfaces of NiO/GaN, HfO₂/GaN, CeO₂/GaN, NiO/Si, HfO₂/Si, and CeO₂/ Si, which need further study.

4. Experimental Section

The n-type Si (100) and the Ga-polar GaN (0001) grown on c-plane sapphire substrates were successively cleaned with acetone, isopropanol, and deionized water. After the solvent cleaning process, the samples were immediately transferred to an ALD chamber for the Al_2O_3 deposition at $300\,^{\circ}\text{C}$. The growth per cycle was $1.2\,\text{Å}\,\text{s}^{-1}$ measured by ellipsometry. For the Kelvin probe force microscopy (KPFM) measurements, 2 nm Al_2O_3 was deposited onto the GaN and Si. 10 nm Al_2O_3 was also deposited onto the GaN and Si samples for scanning transmission electron microscope (STEM) and electron energy loss spectroscopy (EELS) measurements. The STEM sample was fabricated by a focused ion beam (Helios G4, FEI) technique. The EELS spectroscopy was acquired using a Titan G2 60-300 (FEI) operated at 300 kV, equipped with a spherical aberration corrector. The step for each EELS acquisition is 0.8 nm, and the pixel time is 0.02 s.

Density functional theory (DFT) calculations were carried out to analyze the impact of the Al and O defects on the electronic properties of the Al₂O₃/GaN and Al₂O₃/Si heterostructures. DFT calculations were carried out with the Perdew–Burke–Ernzerhof version generalized gradient approximation for the exchange–correction potential. The cutoff energy was set to 450 eV, and a $6 \times 6 \times 1$ k-mesh was used. Geometry relaxations were done until the energy convergence criterion of 0.02 eV Å and the force of 10^{-5} eV was achieved. The θ -phase Al₂O₃ structure was used to build the Al₂O₃/GaN and Al₂O₃/Si heterostructures, as its bandgap and mass density values were close to ALD Al₂O₃, [³⁶] The wurtzite GaN and cubic Si

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phases were adopted in the heterojunctions. Structural optimization was first applied to the separated GaN and Si structures. Next, the $\mathrm{Al_2O_3/GaN}$ and $\mathrm{Al_2O_3/Si}$ heterostructures were built up without and with defects. Pseudohydrogen atoms were adopted to passivate the N dangling bonds at the bottom of the $\mathrm{Al_2O_3/GaN}$ heterostructure and Si dangling bonds at the bottom $\mathrm{Al_2O_3/Si}$ heterostructure thus guaranteeing that all the density of states originated from the defects at the interfaces. For the $\mathrm{Al_2O_3/GaN}$ heterostructure, pseudohydrogen atoms with 0.75 valence electrons were used. For the $\mathrm{Al_2O_3/Si}$ heterostructures, pseudohydrogen atoms with one valence electron were used. $^{[37]}$ Then, the structural optimization was repeated.

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Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Keywords

atomic coordination, interfaces, negative charges, positive charges

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